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## The 2,3-Homopyrrole Ring System

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Summary The 2,3-homopyrrole ring system has been prepared by the copper(I) bromide decomposition of ethyl diazoacetate in the presence of N-methoxycarbonyl-pyrrole.

FURAN AND THIOPHEN behave as dienes toward carbenes and carbenoids giving 1,2-addition products, whereas pyrrole and its derivatives have previously yielded only substitution products.<sup>1</sup> We report what we believe to be the first carbene or carbenoid addition to a pyrrole derivative providing the first synthesis of the 2-azabicyclo-[3,1,0]hex-3-ene ring system (2,3-homopyrrole).

When ethyl diazoacetate was decomposed with copper(1) bromide at  $85^{\circ}$  in the presence of *N*-methoxycarbonylpyrrole (1 equiv.) a complex mixture of products was obtained. T.l.c. of this mixture using silica gel provided a

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14% yield of the mono-adduct (1) and 5% of the bisadduct (2).<sup>†</sup> The n.m.r. spectrum  $(CCl_4)$  of (1) showed  $\tau$  3·47 (d, br, 1H, J 4·5 Hz., H<sub>e</sub>), 4·52–4·70 (m, 1H, H<sub>d</sub>), 5.56-6.08 (m, 3H, H<sub>b</sub> and OCH<sub>2</sub>Me), 6.23 (s, 3H, OCH<sub>3</sub>), 7.12-7.40 (m, 1H, H<sub>c</sub>), 8.75 (t, 3H, J 7.0 Hz., OCH<sub>2</sub>CH<sub>3</sub>), and 9.08 (t, 1H, J 2.0 Hz.,  $H_a$ ). Irradiation of  $H_a$  causes the signal due to  $H_c$  to collapse into a doublet (J 7.0 Hz.) of doublets (J 2.5 Hz.). Irradiation of H<sub>c</sub> causes the H<sub>d</sub> signal to collapse into a doublet (J 4.0 Hz.). The i.r. spectrum (neat) showed important absorptions at 3120, 3060 (vinyl and cyclopropyl hydrogens), 1720 (C=O), and 1590(C=C) cm.<sup>-1</sup> The u.v. spectrum (95%) EtOH) showed  $\lambda_{\max}$  244 ( $\epsilon$  195). The mass spectrum showed a molecular ion at m/e 211 (12%) and the base peak at m/e 138  $(M^+ - CO_2C_2H_5).$ 



The bis-adduct (2) was recrystallized from ether-pentane to give colourless crystals, m.p. 97-98°. The n.m.r. spectrum (CCl<sub>4</sub>) showed  $\tau$  5.83 (q, 4H, J 7.5 Hz., OCH<sub>2</sub>Me), 6.28 (s, 3H, OCH<sub>3</sub>), 6.60 (d broad, 2H, J 7 Hz., H<sub>b</sub>), 7.52— 7.78 (m, 2H, H<sub>c</sub>), 8.17-8.30 (m, 2H, H<sub>a</sub>), and 8.72 (t, 6H, J 7.5 Hz.,  $-OCH_2CH_3$ ). Irradiation of  $H_a$  caused the signal due to  $H_c$  to collapse into a doublet (J 7 Hz.). The i.r. spectrum (KBr) showed important absorptions at 3090, 3063 (cyclopropyl hydrogens) and  $1712 \text{ cm}^{-1}$  (C=O).

The ethoxycarbonyl group would be predicted to occupy the exo-position since this is the stereochemistry of the main adduct from cyclopentadiene and ethyl diazoacetate.<sup>2</sup> The n.m.r. spectrum of (1) also suggests this configuration since the proton  $\alpha$  to the ethoxycarbonyl group (H<sub>a</sub>) is probably shielded by the double bond and occurs at  $\tau$  9.08. This proton (H<sub>a</sub>) appears downfield at approximately  $\tau$  8.23 in the bis-adduct consistent with the above hypothesis and the carbocyclic analogue.<sup>3</sup>

When ethyl diazoacetate is decomposed with copper(I) bromide at 86° in the presence of (1) (1 equiv.) a 10% yield of (2) is obtained. This suggests that the ethoxycarbonyl groups are exo to the five-membered ring in (2). However, the configuration of the cyclopropyl groups with respect to each other cannot be assigned with certainty.

Irradiation of an equimolar mixture of ethyl diazoacetate and N-methoxycarbonylpyrrole using a quartz filter and a 500 w Hanovia high-pressure mercury lamp gave no detectable amount of (1). Irradiation using a Pyrex filter gave only a trace of (1).

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<sup>†</sup> Satisfactory elemental analyses for (1) and (2) have been obtained.

<sup>1</sup> L. A. Paquette, "Principles of Modern Heterocyclic Chemistry," Benjamin, New York, 1968, p. 131; C. W. Rees and C. E. Smithen, Adv. Heterocyclic Chem., 1964, 3, 63. <sup>2</sup> J. Warkentin, E. Singleton, and J. F. Edgar, Canad. J. Chem., 1965, 43, 3456.

<sup>3</sup> H. Dürr, Annalen, 1967, 703, 109.